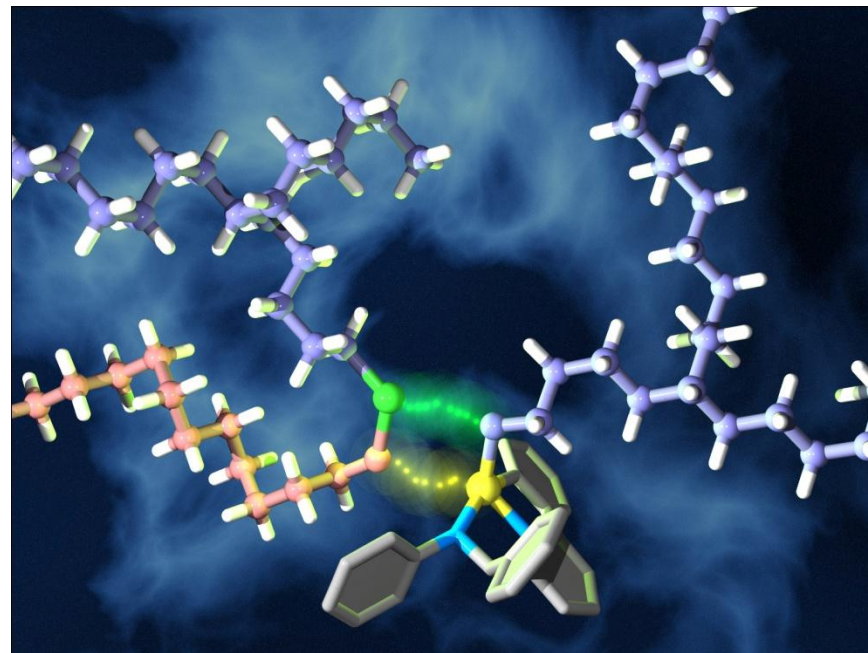


# Performance Polyolefins by Molecular Design

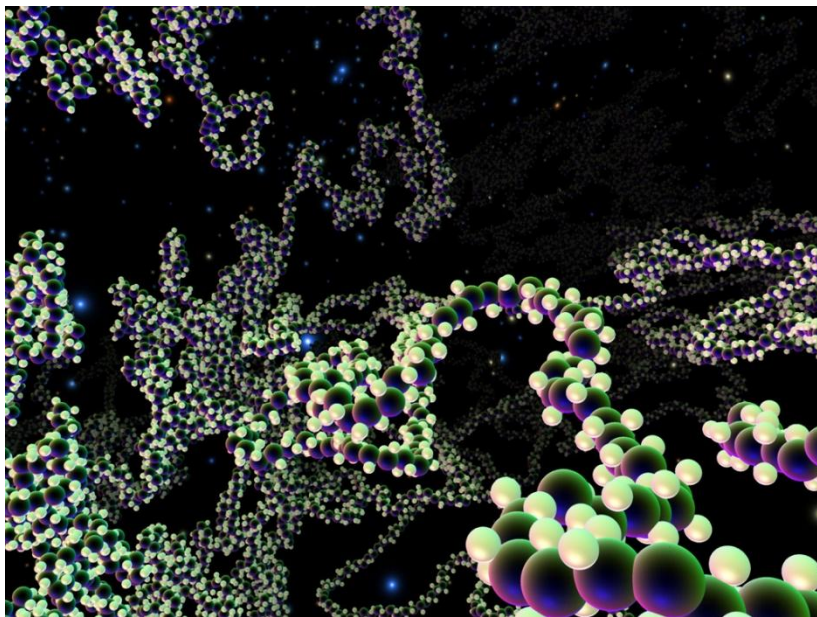
Edmund (Ted) Carnahan  
Fellow, Performance Plastics R&D  
The Dow Chemical Company



# Why Continue to Study Polyolefin Catalysis?

“[W]hen I happen to declare that I keep on studying enthusiastically Ziegler-Natta chemistry, the majority of my colleagues look at me as at an elderly and somehow pathetic gentleman who refuses to admit that the object of his juvenile passion is now withered.”

Vincenzo Busico, *Macromol. Chem. Phys.* **2007**, *208*, 26.



## Customers Like Olefin Solutions

- Polyolefin plastics are inexpensive, non-toxic, light weight, recyclable, and easy to process.
- If an olefin solution meets performance needs, it is usually used.

<http://www.molecularium.com/>

# Polyolefins are Everywhere

- Global annual demand exceeds 120 million metric tons ( $>2.8 \times 10^{11}$  lbs)!
- Growing at ~5% per year, faster growth in emerging economies

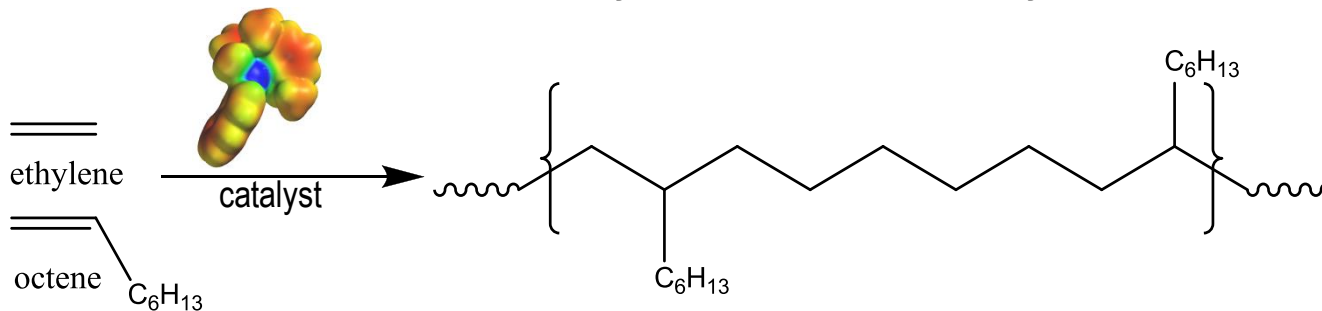


Consumer Durables and Appliances  
Electrical and Electronics  
Food and Specialty Packaging  
Health and Hygiene  
Industrial and Consumer Packaging  
Pipes and Fittings  
Rigid Packaging

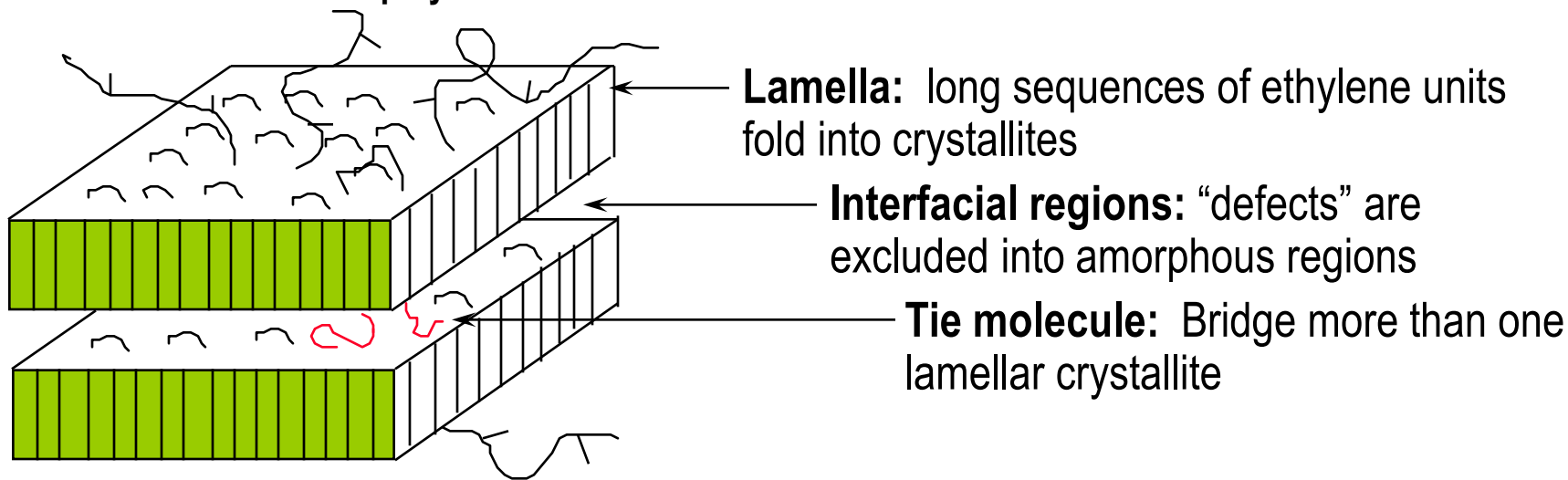


# Polymer Properties Determined by Catalysis

Composition and molecular structure of each polymer chain determined by relative kinetic rates:



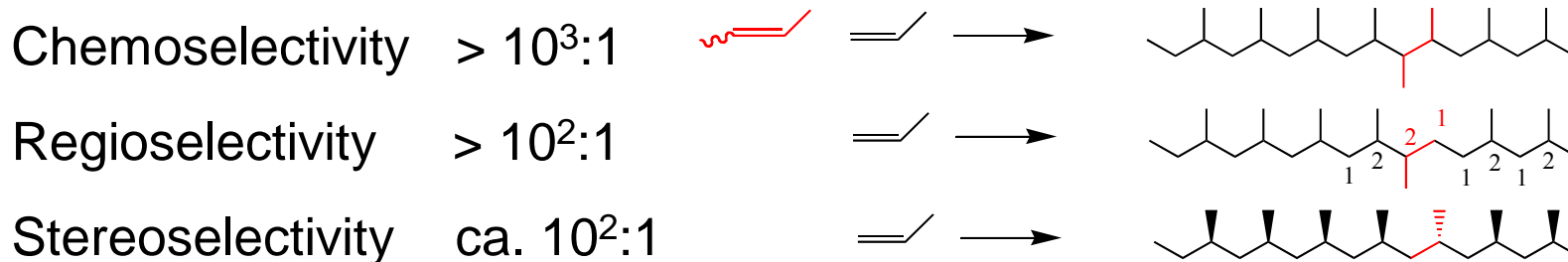
Molecular structure of polymer chains determines bulk structure:



Catalysis  $\rightarrow$  Molecular Structure  $\rightarrow$  Bulk Structure  $\rightarrow$  Properties

# Controlling Microstructure via Catalysis

Polyolefin catalysts are some of the most powerful man-made catalysts invented!



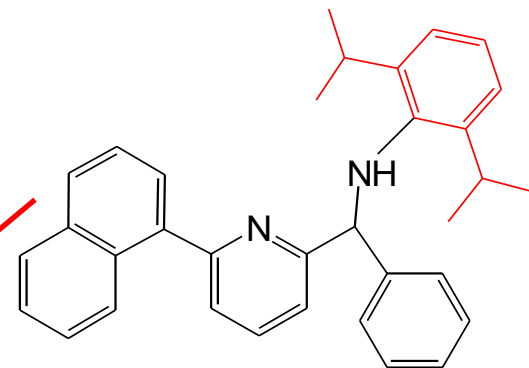
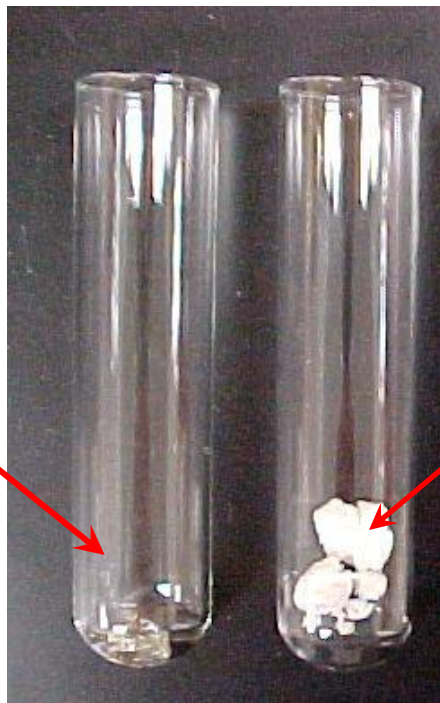
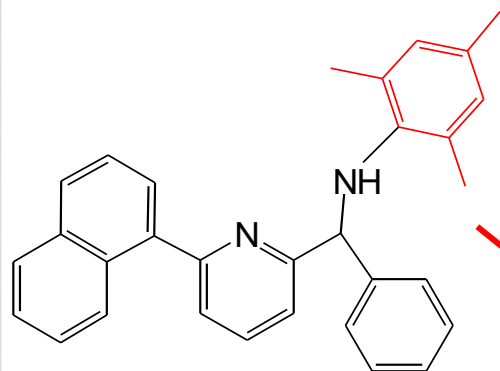
Catalytic rates:  $\sim 10^4 - 10^5 \text{ sec}^{-1}$  (compare to enzymes,  $\sim 10^2 - 10^5 \text{ sec}^{-1}$ )

Productivity: Can be  $> 10^8$  T/O in 10 minutes

Thanks to Prof. Clark Landis

**Control of polymer microstructure requires a mastery of the underlying kinetics**

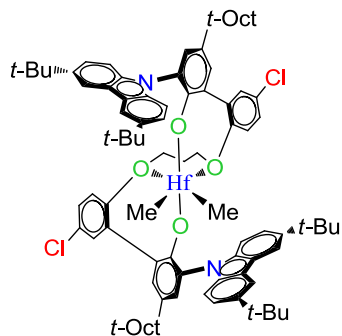
# Minor Changes have Major Implications



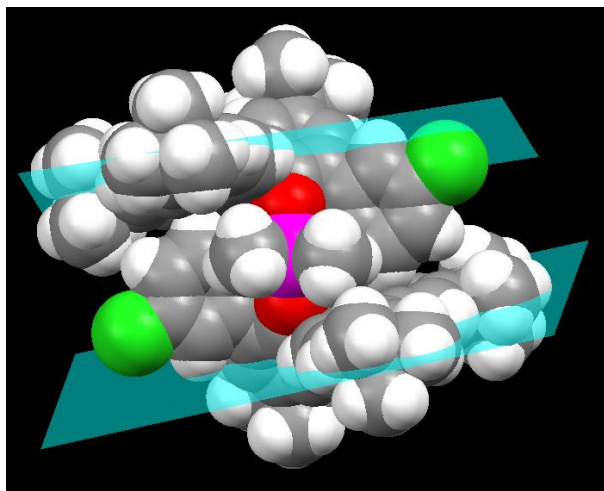
- These two ligands resulted in catalysts which made two very different polymers.
- High Mw, crystalline iPP on the right; low Mw, low tacticity PP on the left.
  - Identification of catalyst structure-function relationships can be greatly accelerated by the use of High Throughput techniques

# Controlling Comonomer Incorporation

Tailor the kinetic reactivity ratio of your catalyst for optimal flexibility.

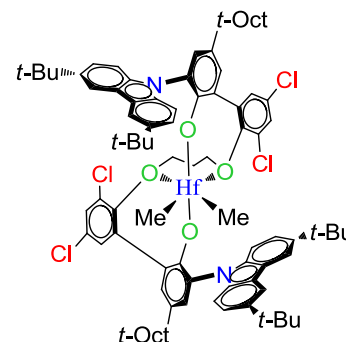


31.3 mol%  
T<sub>m</sub> = -24.5°C

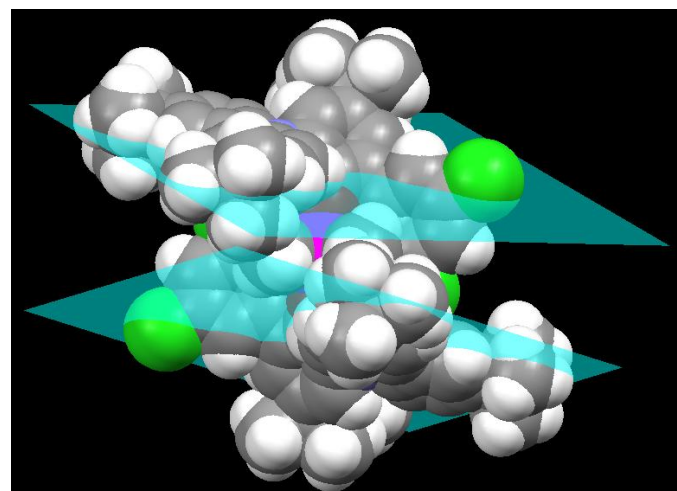


dihedral angle = -19.0°

Batch ethylene-octene  
copolymerization



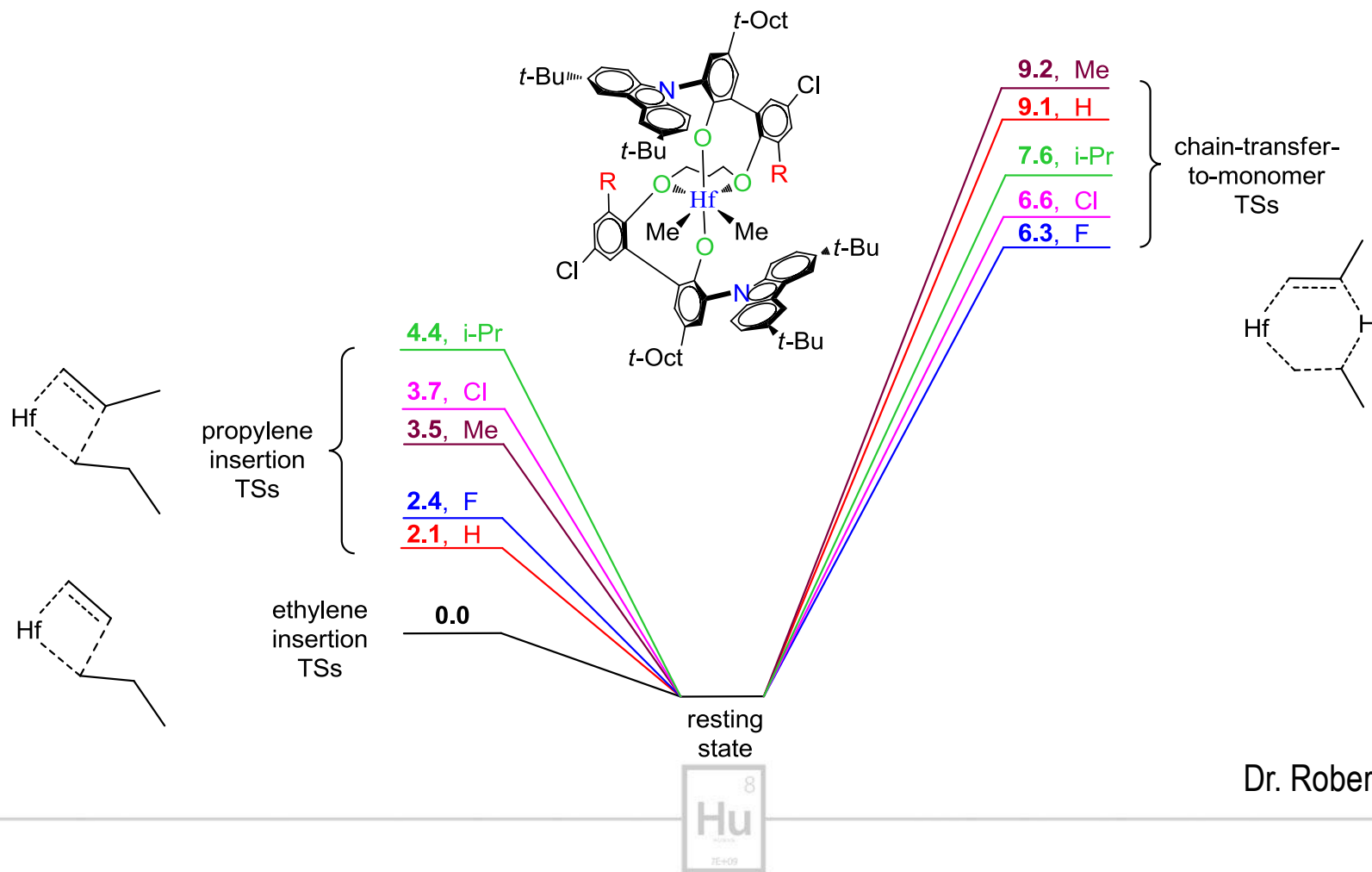
3.5 mol%  
T<sub>m</sub> = 101°C



dihedral angle +28.9°

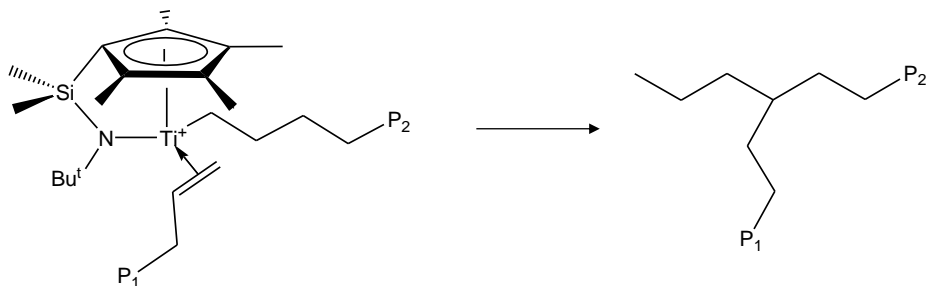
# Calculations can Guide Synthesis

- DFT calculations help guide catalyst families toward the desired  $\alpha$ -Olefin Incorporation and Mw Selectivities



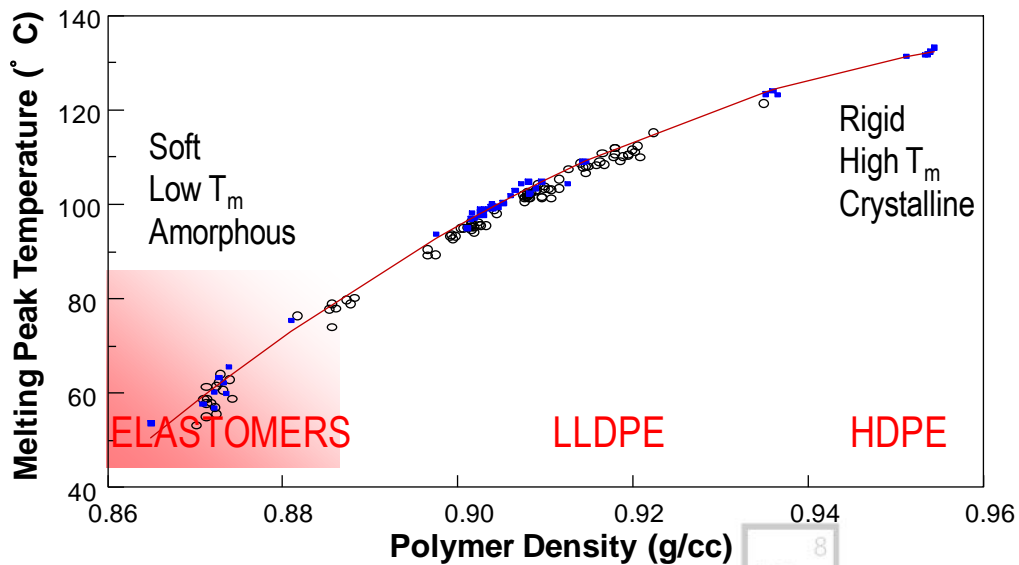
# “Single Site” Catalysts Have Limits

- Even “perfect” kinetic control of a single molecular catalytic site results in structural compromises limiting the applications of your resin.



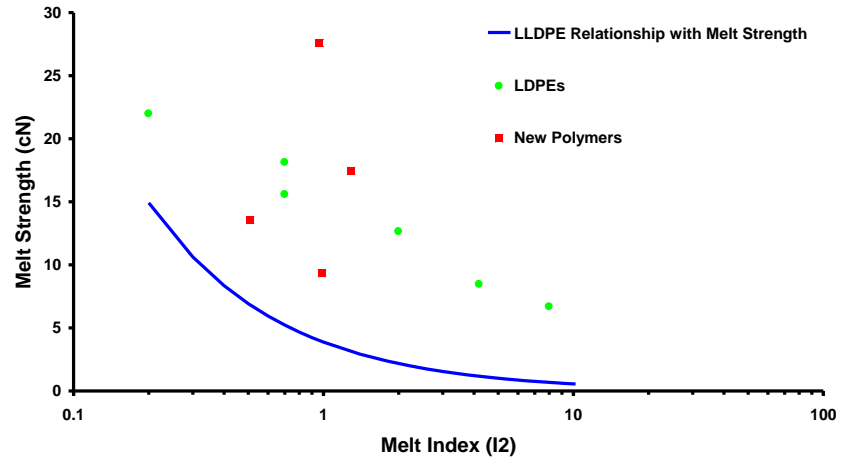
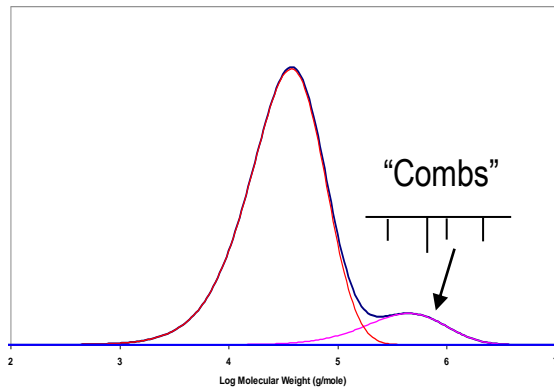
For a single-site catalyst system, the branches have the same Mw as the “backbone,” which can correlate rheology and mechanical properties.

Properties are Correlated



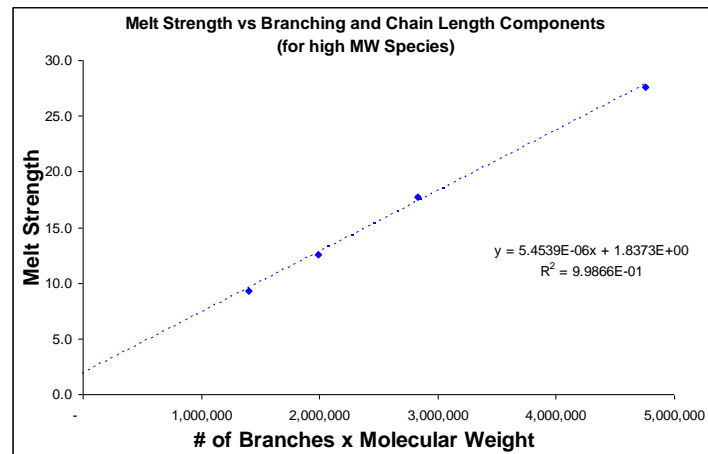
Statistical (e.g. 1<sup>st</sup> order Markovian) comonomer distribution correlates modulus and melting point.

# Control Size and Distribution of LCB

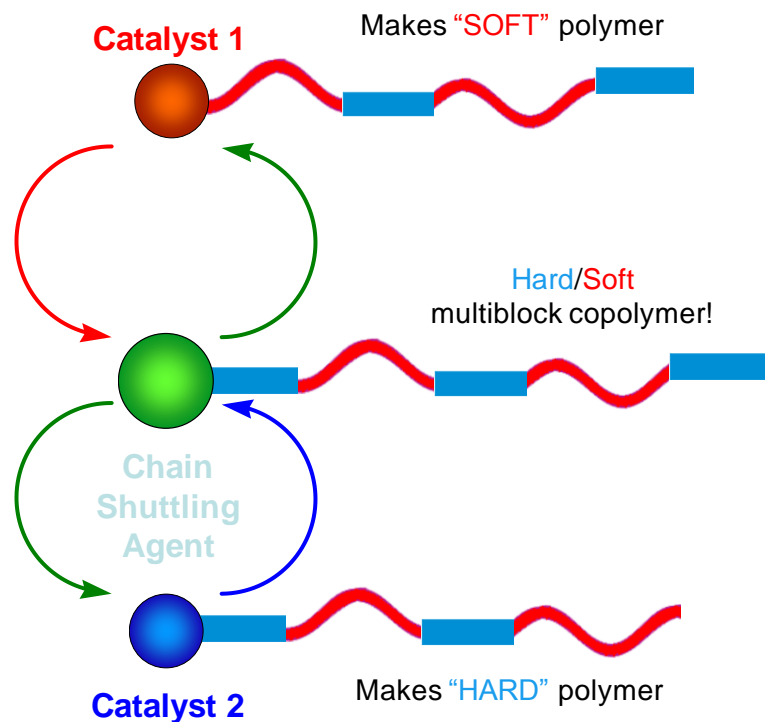


Dual catalyst technology to place comb molecules in the high molecular weight tail.  
 (See, e.g. Dow US patents 7,999,039; 7,300,983; 6,875,816; 6,942,342.)

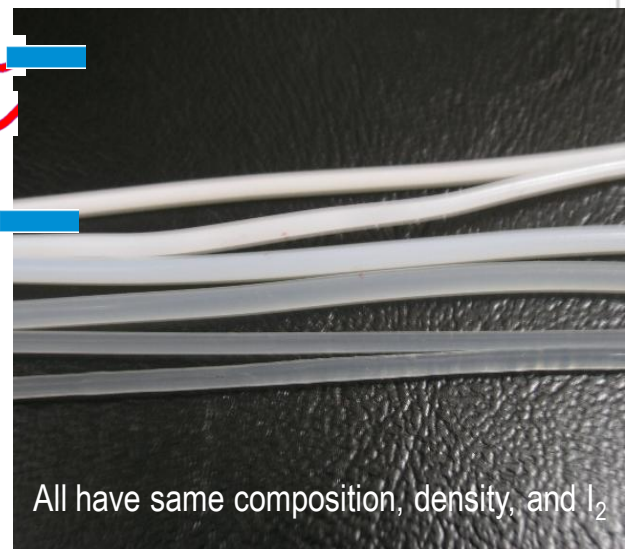
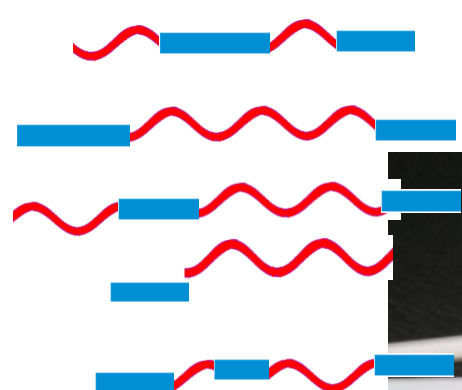
Tandem catalysis can be used to “dial in” melt strength by adjusting the concentration of comb molecules



# OBCs by Chain Shuttling Catalysis



Chain shuttling generates a **HARD/SOFT** multiblock copolymer. Note: Not stoichiometric in CSA!

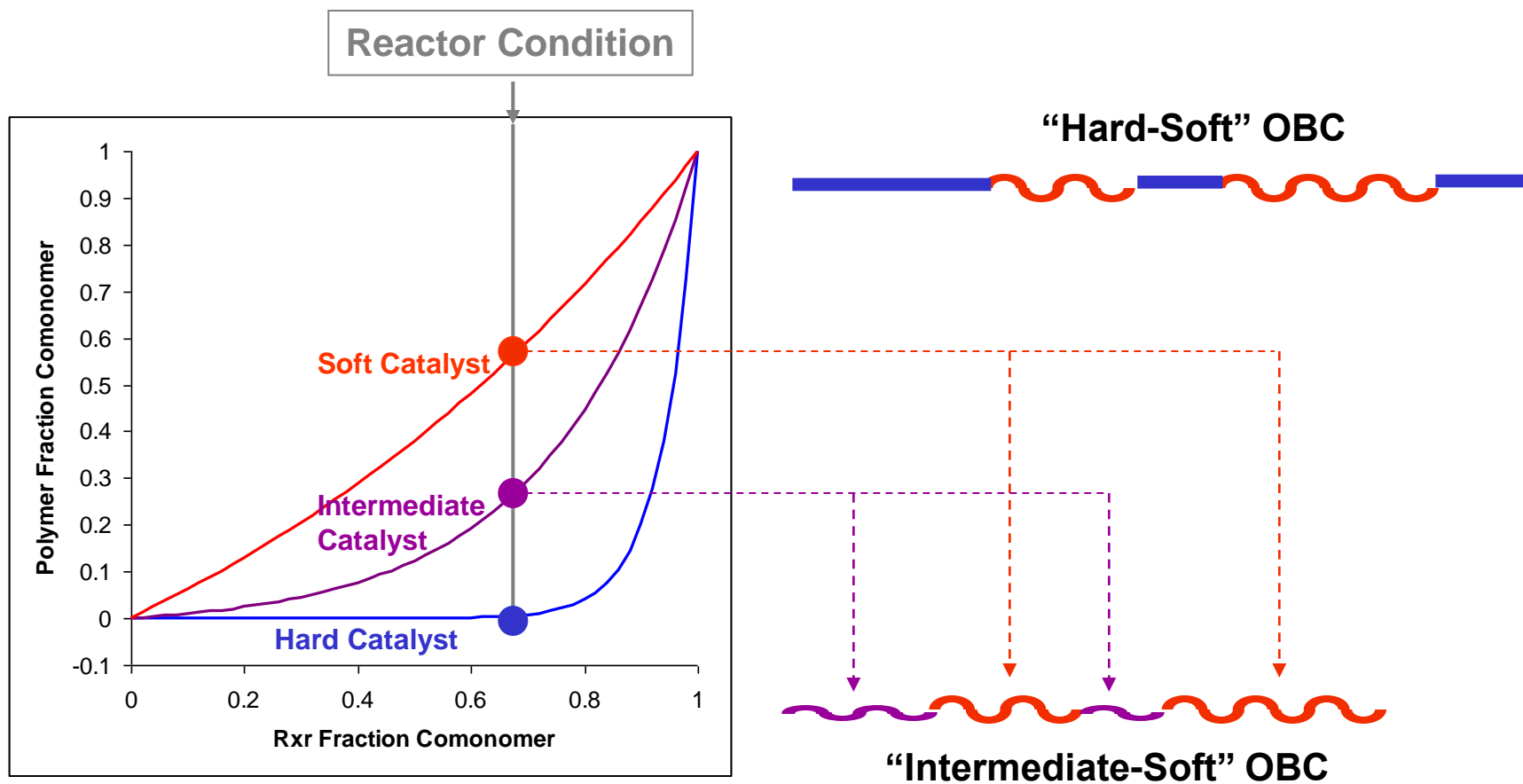


Breaks correlation between modulus and melting point!

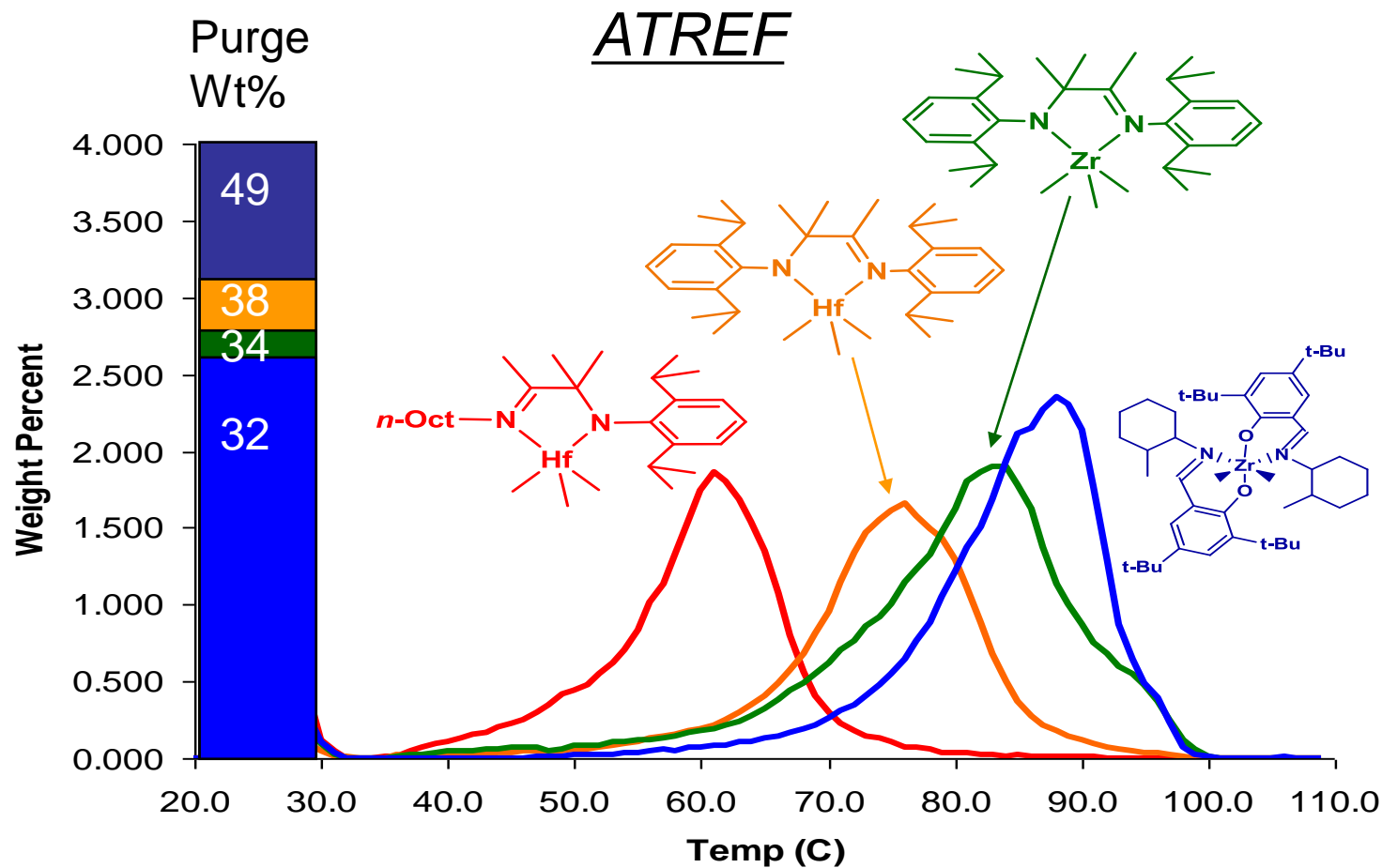
All have same composition, density, and  $I_2$

Arriola, D. J.; Carnahan, E. M.; Hustad, P. D.; Kuhlman, R. L.; Wenzel, T. T. *Science* 5 May 2006: 714.

# Controlling the OBC Density Split



# New Catalysts Can Tune the Comonomer Content of the Hard Block



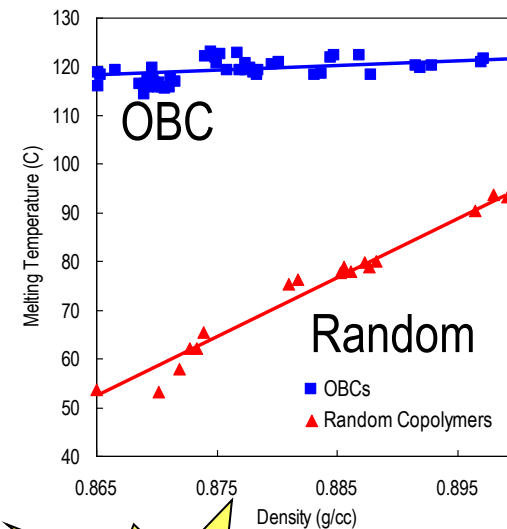
Kuhlman, R. L.; Klosin, J. *Macromolecules*, 2010, 43, 7903.



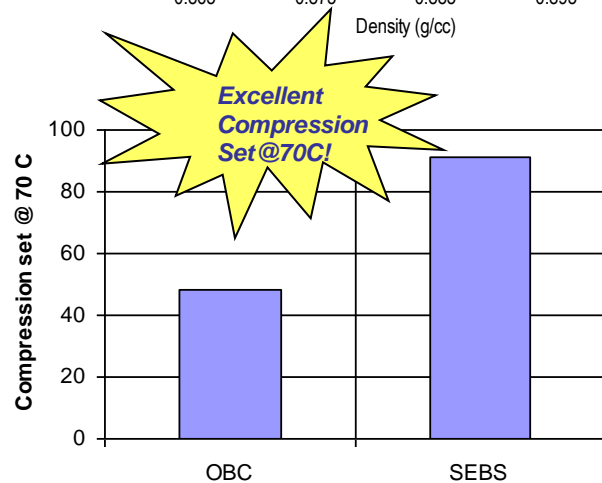
# Commercialized by Dow as INFUSE™ OBCs



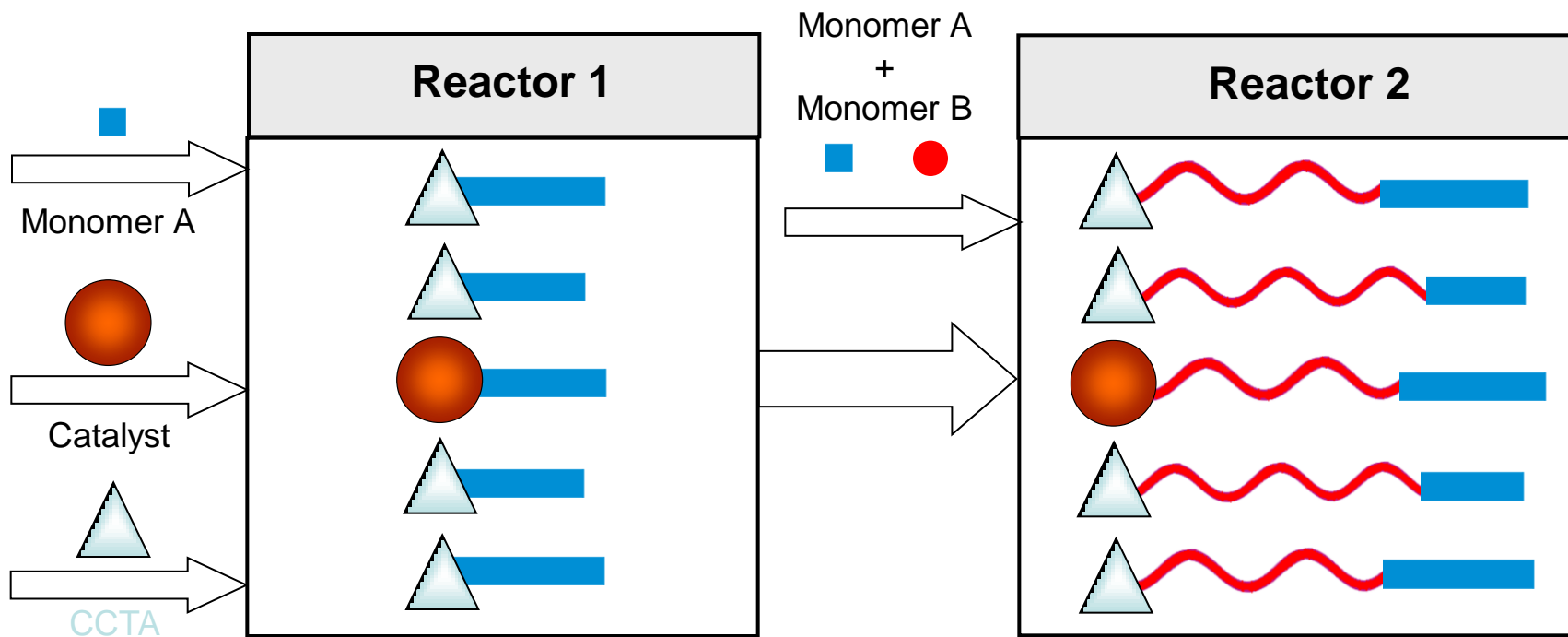
- Outstanding flexibility-high temperature resistance balance
- Fast set-up in processing (shorter cycle time) – soft compounds!



- Excellent elastic recovery properties
- Good compression set performance at room and elevated temperatures



# Dual Reactor CCTP Concept: Sequential Monomer Addition



Coordinative  
chain transfer  
agent

**CCTP generates  
HARD/SOFT diblock!**

# A New Level of Microstructure Control

## Random Copolymer



- Billions of pounds made and sold world-wide
- “Best” materials are statistically random (kinetic control)

- From random copolymers to statistical multiblocks. ✓

## Statistical Multiblock



Hard blocks

- Statistical multiblocks with excellent heat/elasticity balance
- Commercialized as INFUSE™ Olefin Block Copolymers

See: *Science*, **2006**, 312, 714.

- Statistical diblocks by coordinative chain transfer polymerization. ✓

## Statistical Diblock



Hard block

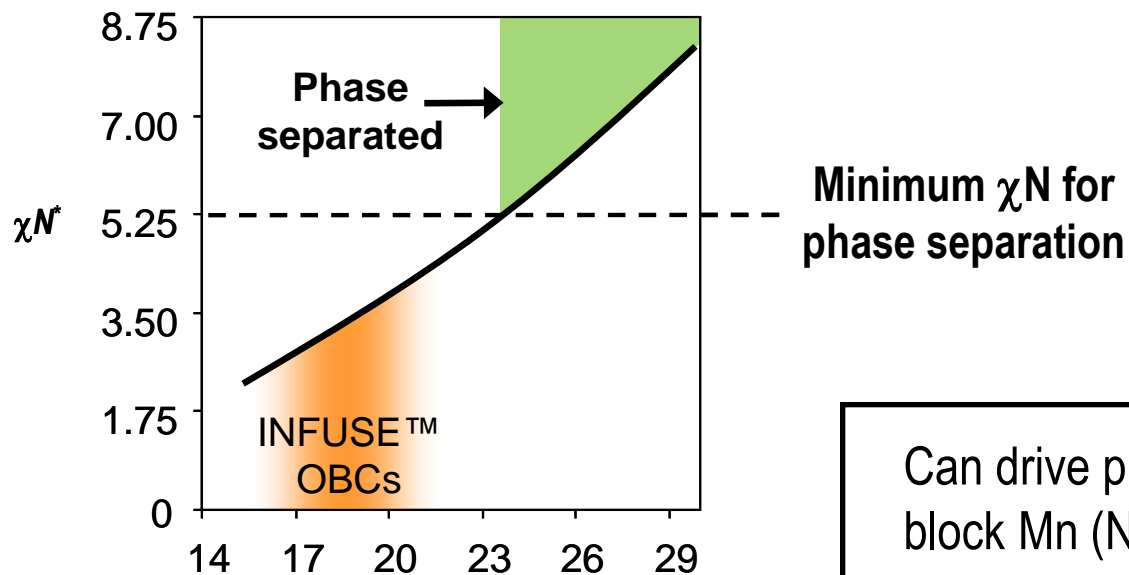
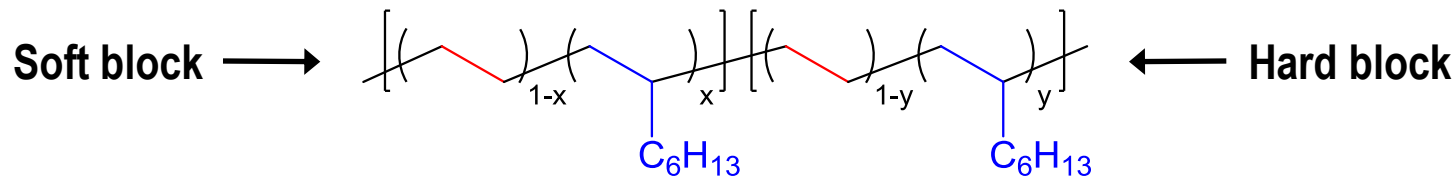
- Exactly two blocks per chain
- Statistical distribution of block lengths

See: *Macromolecules*, **2007**, 40, 7061.

Is this as far as we can push polyethylene performance?

No! Molecular structure can also determine melt-state thermodynamic phase behavior.

# Design OBCs to Phase Separate in the Melt



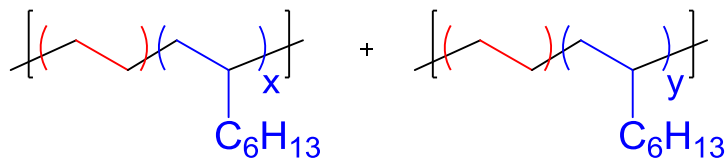
$N^*$ :  $M_n = 40 \text{ kg/mol}$   $\Delta$  mole% octene  
**50% hard segment**

Modeling: Jeff Weinhold

Can drive phase separation by increasing block  $M_n$  ( $N^*$ ) or incompatibility ( $\chi$ ).

To avoid processing issues, we chose to increase octene in the soft block.

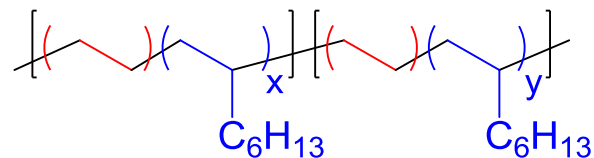
# Ordered Morphologies Observed



Physical Blend

$M_n = 45 \text{ kg/mol}$ , PDI = 4

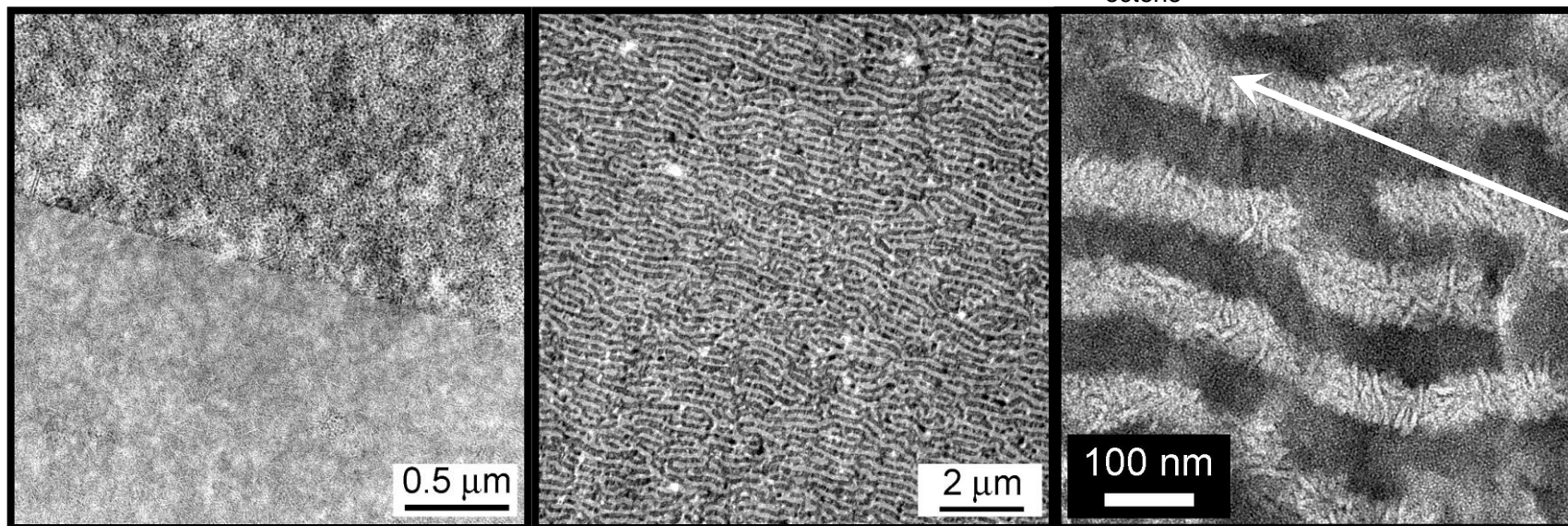
$\Delta_{\text{octene}} = 40$



High Octene OBC

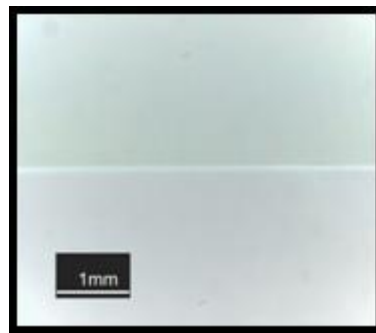
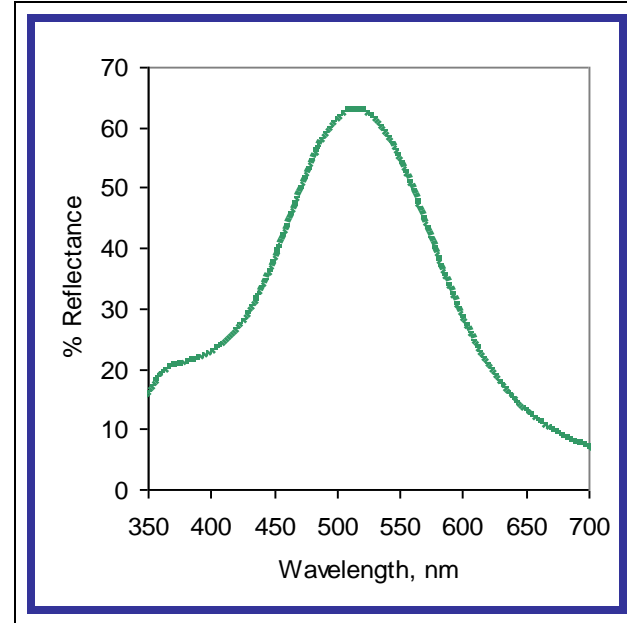
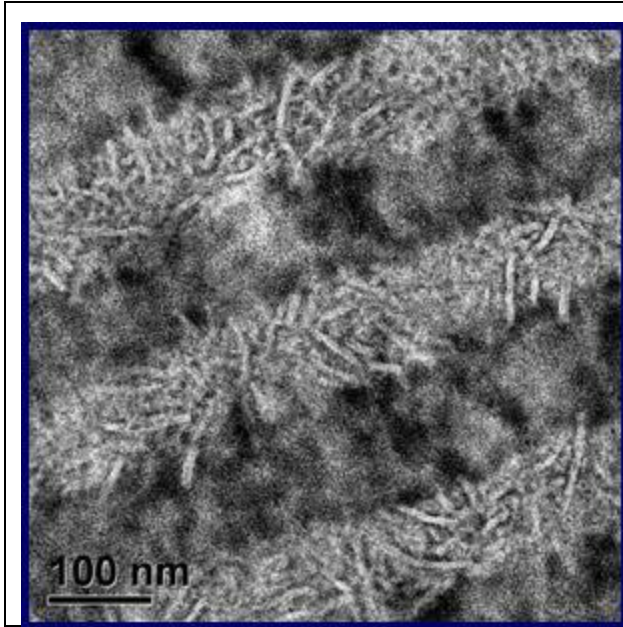
$M_n = 47 \text{ kg/mol}$ , PDI = 2

$\Delta_{\text{octene}} = 36$

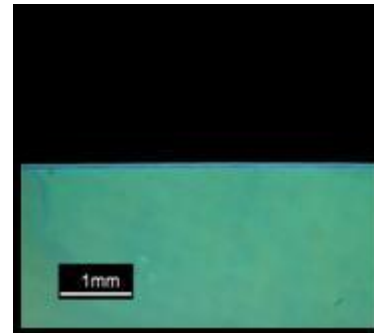


Hustad, Marchand, Garcia-Meitin, Roberts, Weinhold *Macromolecules* **2009**, 42, 3788

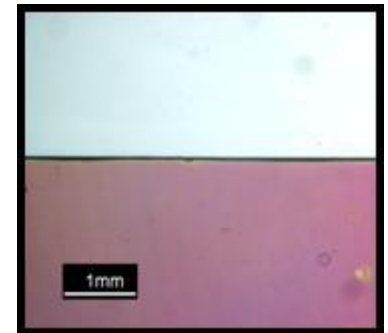
# Simple Pressed Plaques Scatter Blue Light



*Reflected light  
on solid white plate*



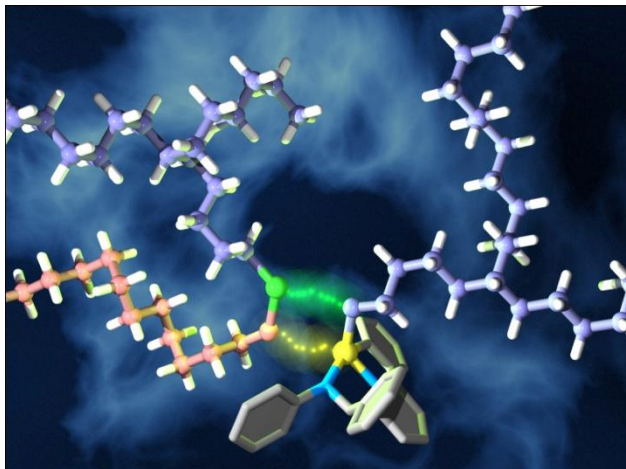
*Reflected light  
on solid black plate*



*Transmitted Light  
thru clear glass plate*



# Innovation in Polyolefins Requires a New Level of Microstructural Control



Mastery of catalyst structure-function relationships to tune polymer properties is critical.

Achieving new combinations of properties from the same mixture of olefins requires even more elegant catalyst solutions.

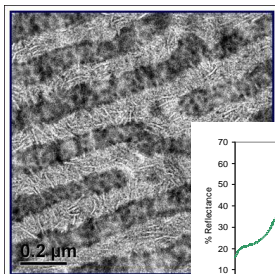


Statistical multiblocks

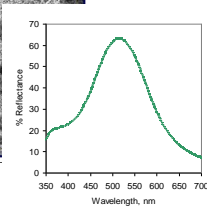


Hard block

Statistical diblocks



Olefin-based advanced  
Optical materials



...and beyond.



# Acknowledgments

## Core R&D

Dan Arriola

Rob Froese

Phil Hustad

Roger Kuhlman

Jeff Weinhold

Tim Wenzel

## Analytical

Eddy Garcia-Meitin

## Dow Elastomers

Tom Karjala

Colin Li Pi Shan

Gary Marchand

Kim Walton

## Pilot Plants

Jan Bazen

Curvel Hypolite

Christine Westgate

... and many more.

